

# Coupling transport with geochemistry Applications to CO<sub>2</sub> sequestration

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**SHP**CO<sub>2</sub>

**MoMaS**

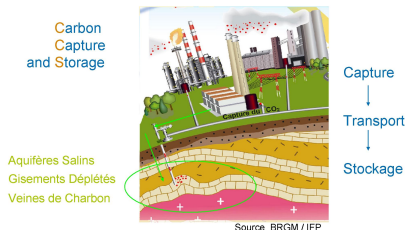
- Motivations and problem statement
- Formulations and solution methods
  - Coupled problem and coupling algorithms
  - A simplified model, and Jacobian preconditioning
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  - MoMaS benchmark
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## Multi-species reactive transport

- Chemistry with equilibrium reactions
- Transport of aqueous species
- Large nonlinear system

Large scale geologic model for CO<sub>2</sub> sequestration project (A. Michel, IFP)



## Numerical issues

- Formulation of coupled problem
- Iterative method: Fixed point vs Newton(-Krylov)
- Preconditioning for Newton-Krylov

# Modeling chemical equilibrium systems

$$\sum_{j=1}^{N_s + \bar{N}_s} \nu_{ij} Y_j \rightleftharpoons 0, \quad i = 1, \dots, N_r \quad N_s + \bar{N}_s \text{ species}, N_r \text{ reactions.}$$

Mass action law  $\nu \log \begin{pmatrix} c \\ \bar{c} \end{pmatrix} + \log K = 0$

Mass conservation  $\mathbf{v}^T \begin{pmatrix} c \\ \bar{c} \end{pmatrix} = \begin{pmatrix} T \\ W \end{pmatrix}$

System of non-linear equations  
 $T$  known from transport,  $W$   
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Given totals  $T$  (and  $W$ , known), split into mobile ( $c$ ) and immobile ( $\bar{c}$ ) total concentrations

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## Chemistry solver

$$\bar{c} = \Psi_c(T, W)$$

## Mobile species

$$\omega \partial_t c_i + \underbrace{\nabla \cdot (\mathbf{u} c_i - \mathbf{D} \nabla c_i)}_{L(c)} = \sum_{j=1}^{N_R} \nu_{ij} R_j, \quad i = 1, \dots, N_S$$

- $c_i$ : concentration of  $i$ th species [mol/l]
- $\mathbf{D}$  Dispersion – diffusion tensor [ $\text{m}^2/\text{s}$ ]
- $R_j$  reaction term (unknown)
- $\omega$ : porosity [–]
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## Mass balance for immobile species

$$\rho_s \partial_t \bar{c}_i = \sum_{j=1}^{N_R} \nu_{ij} R_j, \quad i = 1, \dots, \bar{N}_S$$

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# Formulations and solution strategies

Eliminate **unknown** equilibrium reaction rates by introducing **mobile** and **immobile** totals.

$$\phi \partial_t C + \partial_t \bar{C} + LC = 0$$
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**Others**

- Formulation as DAE (Erhel, de Dieuleveutl, Sabit)
- Elimination technique (Knabner, Kraütle, Hoffmann)
- Nonlinear CG (Bouillard, Herbin, Montarnal)

De Dieuleveut, Erhel, MK (JCP '09)

# A global method from the fixed–point formulation

## Discrete non-linear system

$$\begin{cases} \mathbf{c}^{n+1} = \Psi_T \left( \phi \frac{\bar{\mathbf{c}}^n - \bar{\mathbf{c}}^{n+1}}{\Delta t}, \mathbf{c}^n \right) \\ \bar{\mathbf{c}}^{n+1} = \Psi_C(\mathbf{c}^{n+1} + \bar{\mathbf{c}}^{n+1}) \end{cases}$$

uncoupled



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## Solution by **Newton–Krylov** method

- Solve the linear system by an **iterative** method (GMRES)
- Requires only Jacobian matrix by vector products, Jacobian not stored
- Keep transport and chemistry as black–boxes
- Theoretical framework: inexact Newton's method (Eisenstat and Walker)

Hammond, Valocchi, Lichtner (05), L. Amir's thesis, Amir, MK (09)

# A simplified model: single species with explicit sorption

$c$  mobile concentration,  $\bar{c}$  fixed concentration.

$$\begin{cases} \omega \partial_t c + (1 - \omega) \rho_S \partial_t \bar{c} - \nabla \cdot (\mathbf{D} \nabla c - qc) = 0 \\ \bar{c} = \Psi(c) = \frac{k_f \sigma c}{k_b + k_f c} \quad (\text{Langmuir isotherm}) \end{cases}$$

Structure similar to multicomponent system

Barrett and Kanbner, Knabner and Van Duijn, Frolkovič, Kačur et al.

## 3 formulations

$$\Psi_C(\mathbf{c}) = (\Psi_C(\mathbf{c}_T))_T, T \in \mathcal{T}_h$$

Coupled system  $F_c \begin{pmatrix} \mathbf{c} \\ \bar{c} \end{pmatrix} := \begin{pmatrix} (\mathbf{M} + \Delta t \mathbf{L}) \mathbf{c} + \mathbf{M} \bar{c} - \mathbf{b}^n \\ \bar{c} - \Psi_C(\mathbf{c}) \end{pmatrix} = 0$

Elimination of  $\bar{c}$   $F(\mathbf{c}) := (\mathbf{M} + \Delta t \mathbf{L}) \mathbf{c} + \mathbf{M} \Psi_C(\mathbf{c}) - \mathbf{b}^n = 0$

Elimination of  $c$   $\tilde{F}(\bar{c}) := \bar{c} - \Psi_C((\mathbf{M} + \Delta t \mathbf{L})^{-1}(\mathbf{b}^n - \mathbf{M} \bar{c})) = 0$

# Jacobian preconditioning

- Essential for good linear performance
- Difficult for matrix free formulation
- Only block preconditioning, respect structure of coupled system

Algebraic elimination of mobile concentration **equivalent** to Schur complement of block Gauss–Seidel precondition.

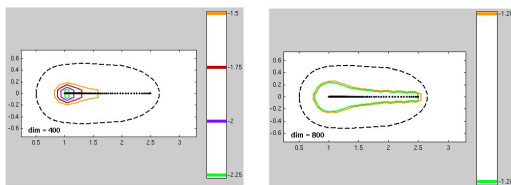
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Can show eigenvalues of preconditioned op. bounded away from 0, **independent of  $h$** , but convergence of GMRES **not** determined by eigenvalues

Field of values analysis ?



Eigenvalues, field of values and pseudospectrum for GS preconditioning

# Preconditioner performance

1D model (Matlab + Sundials),  $h = 0.05$ ,  $K_L = 1.$ ,  $\sigma = 1.5$ , and  $\Delta t = 0.0135$ .

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|      | NI  | LI  | NI    | LI  | NI    | LI  | NI    | LI  |
| None | 3   | 104 | 3     | 167 | 3     | 267 | 3     | 453 |

Mesh dependance : **constant** forcing term

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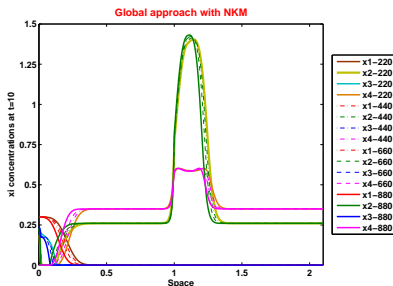
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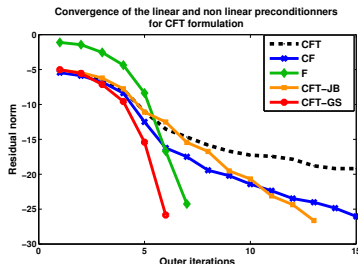
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# The 1D MoMaS benchmark

- 4 aqueous, 1 sorbed primary, 5 aqueous, 2 sorbed secondary species.
- Huge variation in equilibrium constants, large stoichiometric coeffs.
- Long simulation time (not this talk !)

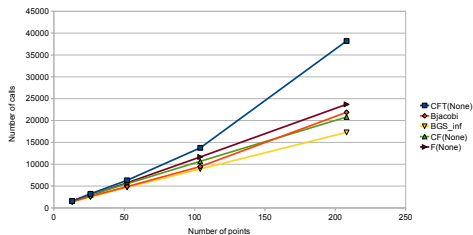


Components concentrations  $t = 10$

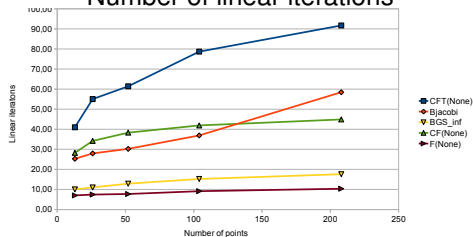


Performance of Newton-Krylov for different preconditioner

## Number of calls to nonlinear function



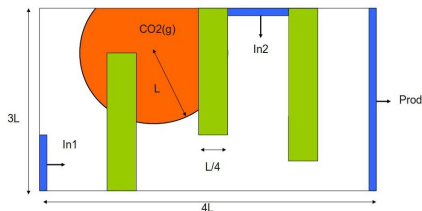
## Number of linear iterations



- Basic method (CFT) not  $h$  independent, other methods OK
- Elimination method has least number of linear iterations, but not least number of  $F$  calls (why ?)
- Block Gauss-Seidel most efficient as preconditioners

# CO<sub>2</sub> sequestration: a synthetic model

Minimal chemical system that still "looks" realistic for CO<sub>2</sub> storage



Dissolution of CO<sub>2</sub> in water, dissolution of calcite. Gas assumed **immobile** (capillary trapping), decouples flow from reactive transport.

## Chemical system

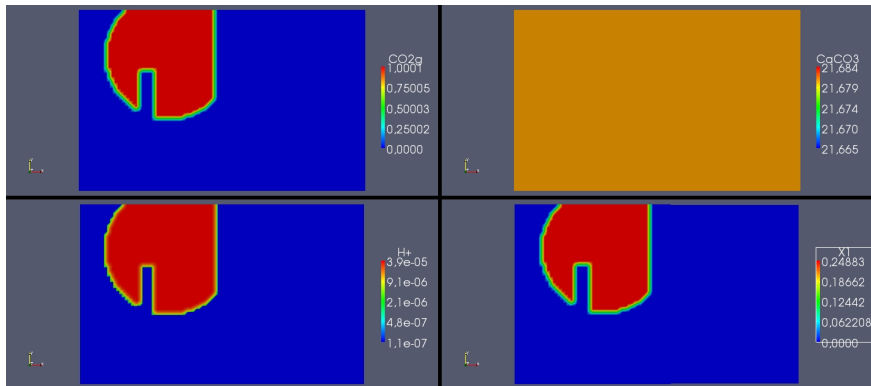
- $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  water dissociation
- $\text{H}_2\text{O} + \text{CO}_2(\text{aq}) \rightleftharpoons \text{HCO}_3^- + \text{H}^+$  dissociation of aqueous CO<sub>2</sub>
- $\text{CO}_2(\text{g}) \rightleftharpoons \text{CO}_2(\text{aq})$  gas dissolution
- $\text{CaCO}_3 + \text{H}^+ \rightleftharpoons \text{Ca}_2^+ + \text{HCO}_3^-$  Dissolution of calcite

Simulation code LifeV (EPFL, MOX, INRIA), nonlinear solver Kinsol (LLNL)



# Evolution of concentrations

Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.

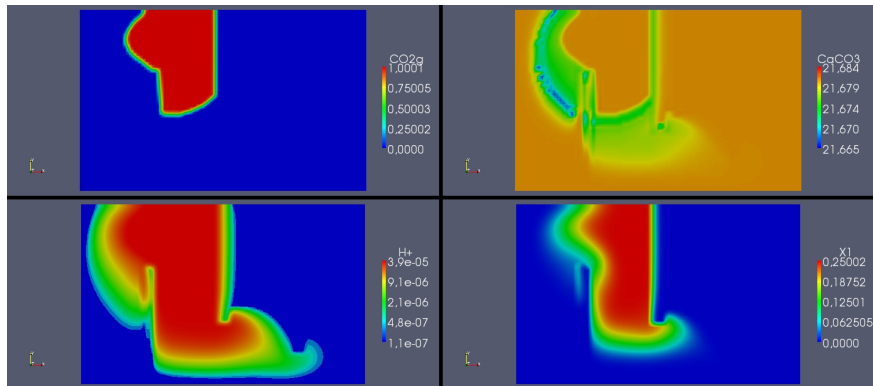


$t = 0$



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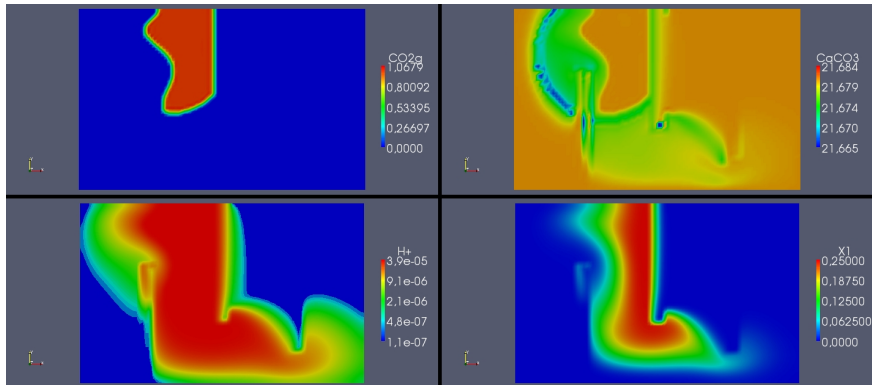
Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



$t = 400$  years

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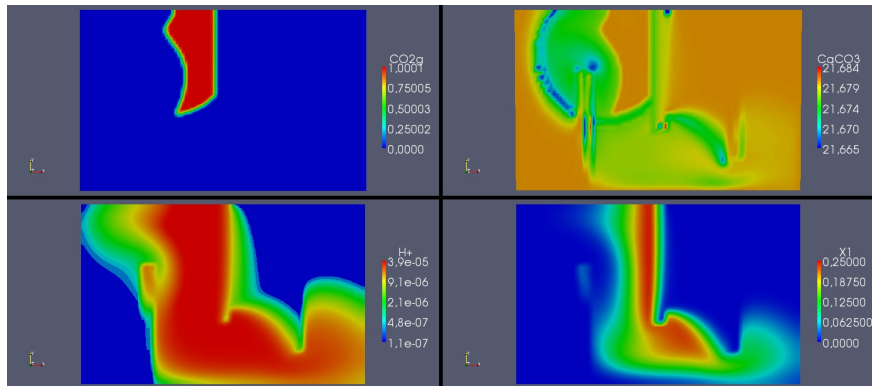
Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



$t = 800$  years

# Evolution of concentrations

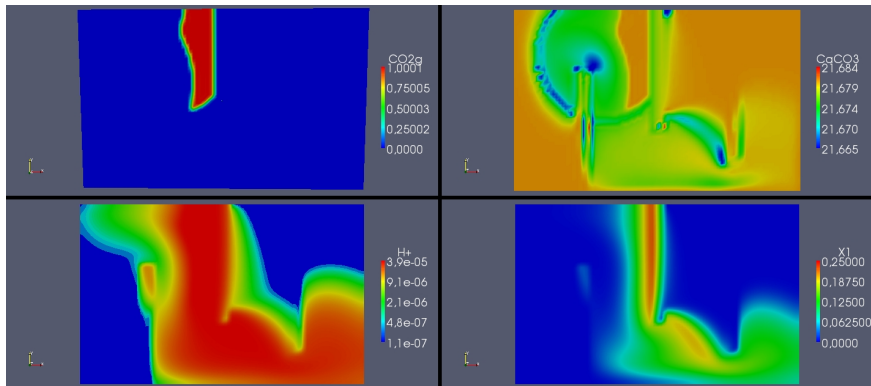
Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



$t = 1200$  years

# Evolution of concentrations

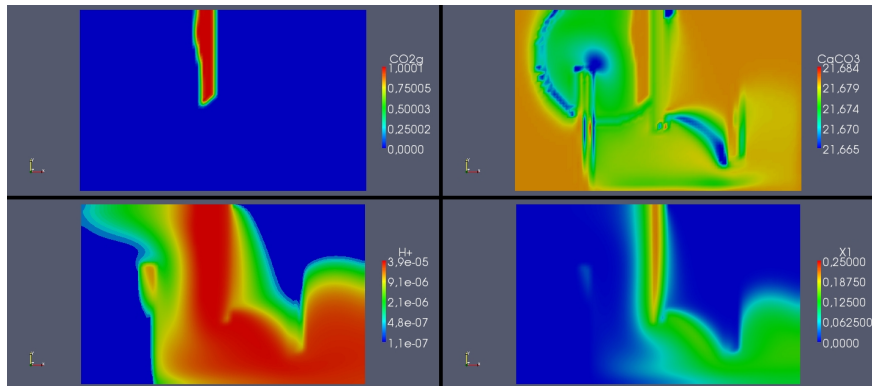
Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



$t = 1600$  years

# Evolution of concentrations

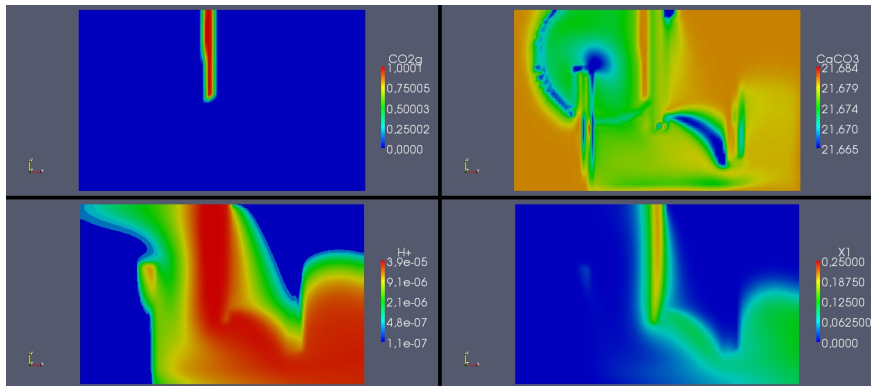
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$t = 2000$  years

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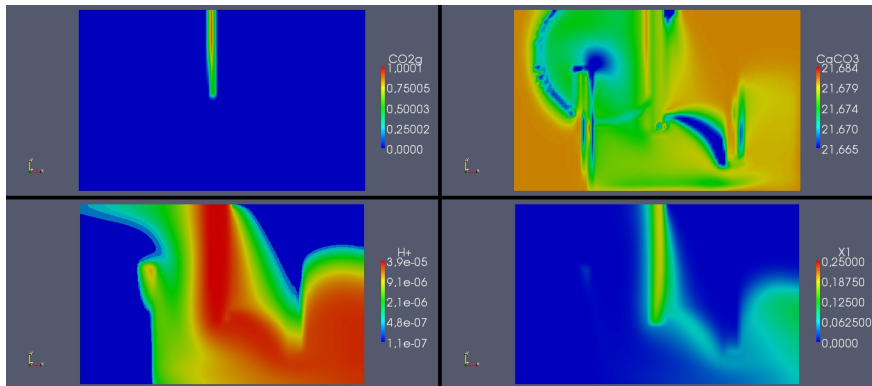
Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



$t = 2400$  years

# Evolution of concentrations

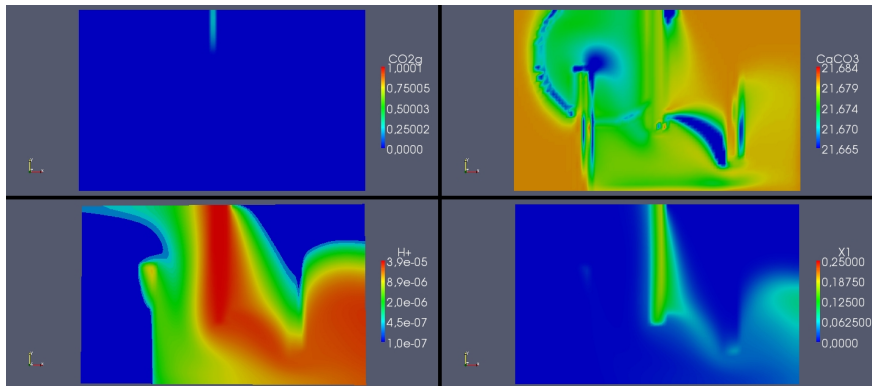
Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



$t = 2800$  years

# Evolution of concentrations

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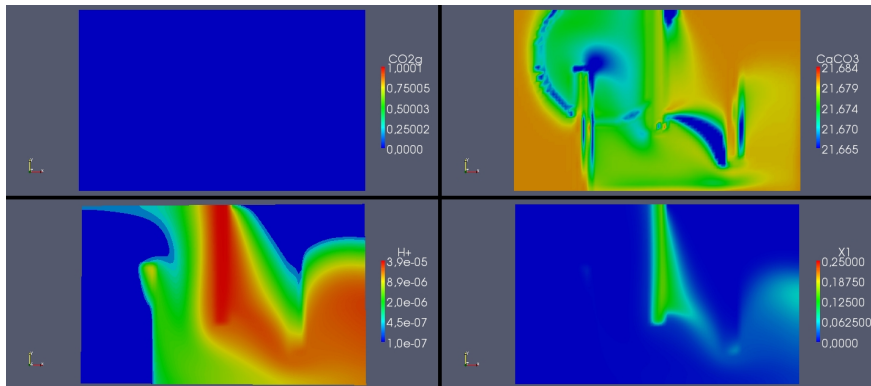


$t = 3200$  years



# Evolution of concentrations

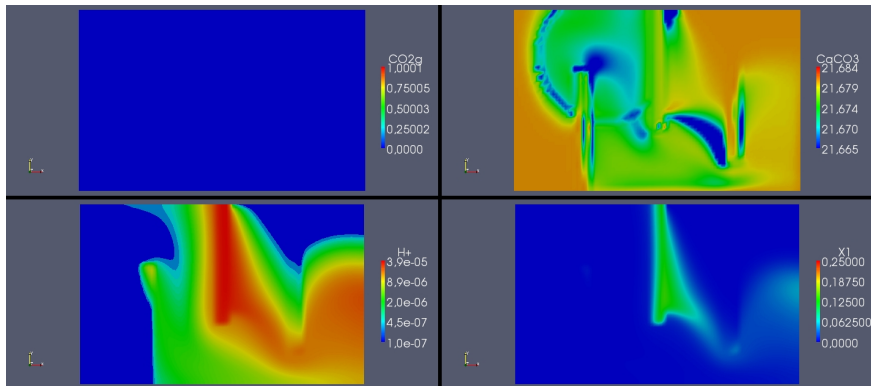
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$t = 3600$  years

# Evolution of concentrations

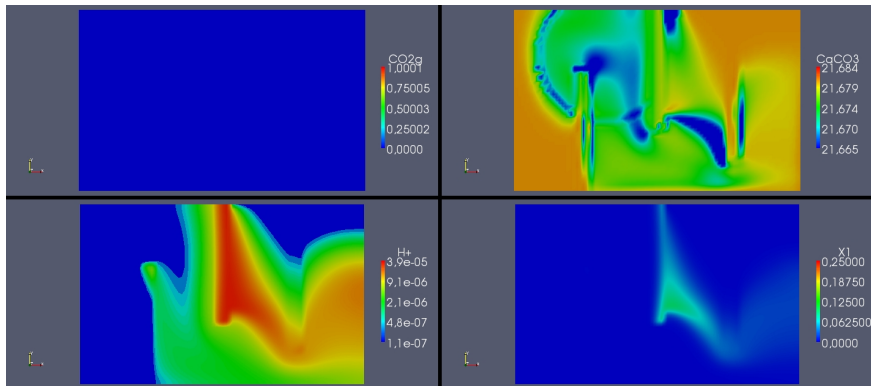
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$t = 4000$  years

# Evolution of concentrations

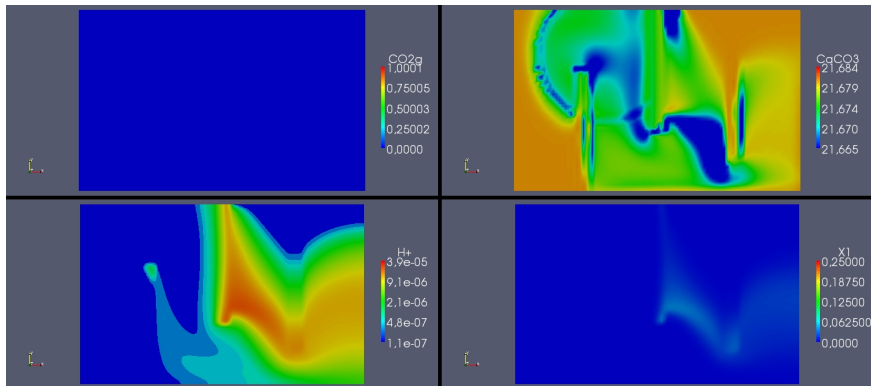
Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



$t = 5000$  years

# Evolution of concentrations

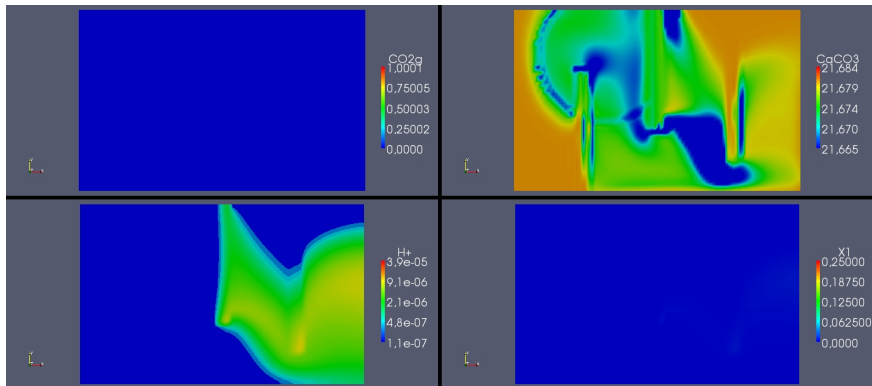
Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



$t = 7000$  years

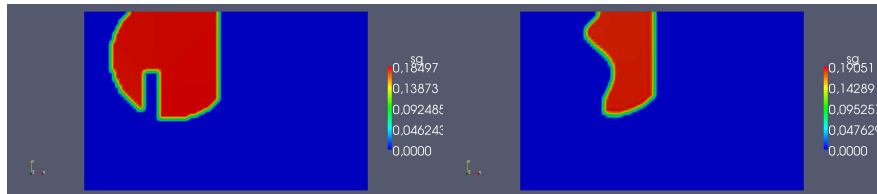
# Evolution of concentrations

Grid with 5700 cells,  $\Delta x = 50\text{m}$ ,  $\Delta t = 100$  years, simulate 10 000 years.



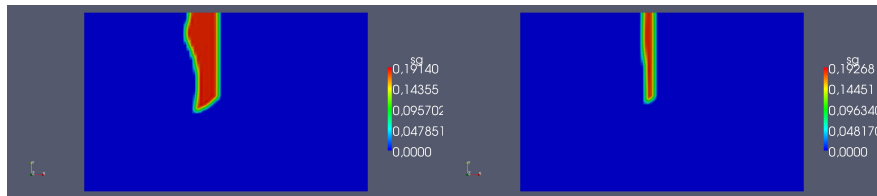
$t = 10000$  years

# Evolution of gas saturation



$t = 0$

$t = 800$  years



$t = 1600$

$t = 2400$  years

*Inria*

- Formulation for **coupled** problem, separate transport and chemistry
- Newton–Krylov method can be applied on “fixed–point” formulation
- Connection between block–preconditioning and elimination at non-linear level
- Inverting transport gives **mesh independent convergence** for both linear (LI) and nonlinear (NI) iterations.
- In practice: approximate inverse should give spectral equivalence
- Future work: Prove FOV results, more tests on multicomponent chemistry, extension to two-phase flow



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Dipartimento di Matematica

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